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Effect of the nature of silicon oxide structures on the activity of an alumina-chromium catalyst in the reaction of iso-butane dehydrogenation

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Abstract

The formation of silicon oxide structures in the composition of an alumina-chromium catalyst for the dehydrogenation of iso-butane is studied by means of nitrogen adsorption, XRD analysis, solid-state ^{29}Si NMR spectroscopy, temperature-programmed desorption of ammonia, and UV-Vis and Raman spectroscopy. It is established that 0.5-1.2 wt % silicon was distributed on the catalyst surface in the form of $\text{Si}(\text{OSi})_4$ structures. As the silicon content was increased to 2.2-3.6 wt %, $\text{Si}(\text{OSi})_3(\text{O}-)$ structural elements were present on the surface in addition to $\text{Si}(\text{OSi})_4$. The formation of silicon oxide structures on the catalyst surface was responsible for an increase in the concentration of $\text{Cr}(\text{III})$ ions and a decrease in the surface acidity; the activity and selectivity of the catalysts in the reaction of iso-butane dehydrogenation increased. © Pleiades Publishing, Ltd., 2014.

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Keywords

Alumina-chromium catalyst, Chromium(III) oxide, Dehydrogenation, Silicon